

PIG® Base Encapsulating & Neutralizing Mat Pad

MAT353 Mediumweight, 11" x 12", 25 pads

Absorb, encapsulate and neutralize small spills of bases in one easy step for safer cleanup. Color indication identifies the best time to handle.

- Encapsulate and neutralize common bases with our mediumweight pads; great for cleaning up spills in areas where chemicals are stored, mixed or transported
- Changes color on contact and during neutralization with liquid bases (pH above 7); color change continues until pH becomes neutral and is safe for handling
- Instantly absorbs up to 50% concentrations of potassium hydroxide and other strong chemical bases
- Gelling agents solidify bases during neutralization; prevents them from leaching out of mat



Specifications

Traffic	Not Recommended for Foot Traffic
Static-Dissipative	No
Indoor/Outdoor	Indoor Use
Floating	No
Reusable	No
Wringable	No
Incinerable	Yes
Fluid Absorbed	Bases
Special Feature	Encapsulating & Neutralizing
Color	Cream
Dimensions	11" W x 12" L
Mat Weight	Mediumweight
Absorbency	Up to 80 oz. per box
Absorbency per	Up to 3.2 oz. per pad
Brand	PIG
Category	Encapsulating & Neutralizing Mat
Perforated	No
Sold as	25 pads per box
Weight	3 lbs.

# per Pallet	360
Composition	Skin - Polypropylene Filler - Polymer, Citric Acid, Thymol Blue
UNSPSC	47131901
UV Resistant	No
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Metric Equivalent

Absorbency per	Up to 94.6 mL per pad
Absorbency	Up to 2.4 L per box
Dimensions	27.9cm W x 30.5cm L
Weight	1.4 kg

Technical Information

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This product is subject to export restrictions and cannot be shipped to Canada.

Technical Documents

[Guidelines for PIG Base Encapsulating and Neutralizing Absorbents](#)

[PIG® Base Encapsulating / Neutralizing Absorbents](#)

Disclaimers

HAZ-MAT Notice

To ensure effectiveness and your safety, we recommend that you conduct compatibility and absorption testing of your chemicals with PIG® HAZ-MAT products prior to purchase. If you have any questions or need samples to test, please call Technical Services.

Base Absorbents Notice

Avoid contact with metallic nitrates, cyanides, sulfides and strong oxidizers. Contact with sodium or calcium hypochlorite creates chlorine gas.



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